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Bases. acid. base.
alkaloids. Arrhenius.
sour taste, dissolves
metals, turns blue
litmus paper red,
mor.... bitter taste,
slippery feel, turns red
litmus paper blue,
more.... organic bases
found in plants that
are often poisonous.

_____ definition:
acids are substances
that produce H^+ ions

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-sour. reaction with

metals- many metals

such as Zn and Mg

react with aqueous

solutions of acids to

produce hydrogen

gas. electrical

conductivity- some

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are strong electrolytes, others are weak. indicators- acids cause certain chemical dyes, called indicators, to change color. Chapter 17 chemistry Acid and Bases Flashcards | Quizlet Chapter 17.

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the words in the box
to fill in the blanks in
the paragraphs below.

An acid is a
substance that
produces _____ in

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solution. Acids taste
_____ . Some acids
can produce painful
burns and damage
tissues.

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17. During the
formation of a
solution, energy is
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solubility of solids

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increases as the
temperature of the
solvent INCREASES.

19. A compound that
produces hydronium
ions (H_3O^+) when
dissolved in water is
a(n) ACID. 20. Please
list and describe the
three ways that

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17. During the formation of a solution, energy is **ABSORBED & RELEASED**. 18. The solubility of solids increases as the temperature of the

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solvent INCREASES.

19. A compound that produces hydronium ions (H_3O^+) when dissolved in water is a(n) ACID. 20. Please list and describe the three ways that

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16.1 Acids and
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Chapter 17 Bases: A Brief Review

•Acids taste sour and cause certain dyes to change color. •Bases taste bitter and feel soapy. •Arrhenius concept of acids and bases: •An acid is a substance that, when dissolved in water, increases the concentration of H^+ ions. •Example: HCl is an acid. •An

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Arrhenius base is a substance that, when dissolved in water, increases the concentration of OH^- ions.

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Questions 1) A _____
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Bases
consists of a nitrogen-
containing base and a
sugar. A) nucleoside

B) base pair C)

nucleotide D)

complementary base

E) pyrimidine Answer:

A 2) A _____ consists

of a nitrogen-

containing base, a

sugar, and a

phosphate group.

phosphate group.

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Acids and Bases

Section 118.18.1

Figure 18.1

Rhododendrons

flourish in rich, moist
soil that is moderately
acidic, whereas

semper-vivum,

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Commonly called hen and chicks, grows best in drier, slightly basic soil. Introduction to Acids and Bases-!).)DEADifferent models help describe the behavior of acids and bases.

Chapter 18: Acids and Bases - FCPS

Acids and bases are important components

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in household products, industrial processes, and in environmental and biological systems. As shown in OWL Activity 16.1, many of the items you might find in your home are acids and bases. We begin our study of acids and bases where we left off in Chapter 5, with the

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Arrhenius acid and
base definitions.

CHAPTER 16: Acids and Bases

There are 20 amino acids, but only 4 nucleotide bases in DNA. That means it takes at least 3 nucleotides to code for one amino acid, hence a codon (three amino acids). The

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code is read as an
RNA code, 5' to 3'.

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1.(s)ofagene inDNA.

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