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Physical Chemistry Problems and Solutions several branches of chemistry. In summary, the book is an up to date and readable summary of an important growing area. of analytical chemistry. It is highly recommended to anyone whose interests include the chemistry of aqueous solutions. MICHAEL D. MORRIS University of Michigan Ann A-bor, Michigan Physical Chemistry Problems and Solutions

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Physical Chemistry Problems. ©Mike Lyons 2013. e4 Mathematics for Physical Chemistry the sine and cosine to the appropriate number of digits. (31) 2? rad 360 = 0.54 rad sin(30.5) = 0.5075 sin(31.5) = 0.5225 sin(31) = 0.51 cos(30.5) = 0.86163 cos(31.5) = 0.85264 cos(31) = 0.86 15. Some elementary chemistry textbooks give the value of R, the ideal gas constant, as 0.0821 l atm K⁻¹ mol⁻¹. a.

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